## Chapter 38 Problem 19 <sup>†</sup>

## Given

 $^{35}_{17}Cl$   $^{35}_{19}K$ 

## Solution

a) Compare the number of nucleons.

The number of nucleons is the same for each isotope. Each has a combination of 35 protons and neutrons.

b) Compare the nuclear charge.

The chlorine has 17 protons and, therefore, has a charge of 17 times  $1.6 \times 10^{-19}$  C. The potassium has 19 protons and has a charge of 19 times  $1.6 \times 10^{-19}$  C.

<sup>&</sup>lt;sup>†</sup>Problem from Essential University Physics, Wolfson