Chapter 38 Problem 15 [†]

Given

 $^{35}_{17}Cl\ ^{35}_{19}K$

Solution

a) Compare the number of nucleons.

The number of nucleons is the same for each isotope. Each has a combination of 35 protons and neutrons.

b) Compare the nuclear charge.

The chlorine has 17 protons and, therefore, has a charge of 17 times 1.6×10^{-19} C. The potassium has 19 protons and has a charge of 19 times 1.6×10^{-19} C.

[†]Problem from Essential University Physics, Wolfson