

Chapter 34 Problem 43 †

Solution

a) Find the rate of photon production for a 1.0 kW radio antenna transmitting at 89.5 MHz.

First find the energy per photon

$$E = hf = (6.63 \times 10^{-34} \text{ J} \cdot \text{s})(89.5 \times 10^6 \text{ Hz}) = 5.93 \times 10^{-26} \text{ J}$$

Now use this value as a conversion factor between joules and photons. A watt is a joule per second. Therefore, the photon production rate is

$$P = 1.0 \text{ kW} = \frac{1000 \text{ J}}{\text{s}} \left(\frac{1 \text{ photon}}{5.93 \times 10^{-26} \text{ J}} \right) = 1.69 \times 10^{28} \text{ photons/s}$$

b) Find the rate of photon production for a 1.0 mW laser transmitting at 633 nm.

First find the energy per photon

$$E = hf = \frac{hc}{\lambda} = \frac{(6.63 \times 10^{-34} \text{ J} \cdot \text{s})(3.0 \times 10^8 \text{ m/s})}{633 \times 10^{-9} \text{ m}} = 3.14 \times 10^{-19} \text{ J}$$

Now use this value as a conversion factor between joules and photons. The photon production rate is

$$P = 1.0 \text{ mW} = \frac{0.0010 \text{ J}}{\text{s}} \left(\frac{1 \text{ photon}}{3.14 \times 10^{-19} \text{ J}} \right) = 3.18 \times 10^{15} \text{ photons/s}$$

c) Find the rate of photon production for a 2.5 kW X-ray machine transmitting at 0.10 nm.

First find the energy per photon

$$E = hf = \frac{hc}{\lambda} = \frac{(6.63 \times 10^{-34} \text{ J} \cdot \text{s})(3.0 \times 10^8 \text{ m/s})}{0.10 \times 10^{-9} \text{ m}} = 1.99 \times 10^{-15} \text{ J}$$

Now use this value as a conversion factor between joules and photons. The photon production rate is

$$P = 2.5 \text{ kW} = \frac{2500 \text{ J}}{\text{s}} \left(\frac{1 \text{ photon}}{1.99 \times 10^{-15} \text{ J}} \right) = 1.26 \times 10^{18} \text{ photons/s}$$

†Problem from Essential University Physics, Wolfson